Tetrahedron Letters,Vol.25,No.23,pp 2475-2478,1984 0040-4039/84 \$3.00 + .00 Printed in Great Britain ©1984 Pergamon Press Ltd.

> KERAMADINE, A NOVEL ANTAGONIST OF SEROTONERGIC RECEPTORS ISOLATED FROM THE OKINAWAN SEA SPONGE AGELAS SP.<sup>1</sup>

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Summary: A new bromine-containing alkaloid, keramadine, has been isolated from the Okinawan sea sponge Agelas sp. as a novel antagonist of serotonergic receptors and its structure has been elucidated by n.m.r. spectrometry.

During our study on physiologically active substances of marine organisms, we have examined the pharmacological actions of 70% ethanolic extracts of numerous sea sponges collected at Okinawa using the isolated vascular smooth muscle. As a result, it was found that a brown sea sponge <u>Agelas</u> sp. had antagonistic activities<sup>2</sup> on serotonergic receptors in the rabbit aorta<sup>3</sup>. In this communication we report the isolation and structure determination of a new antagonist of serotonergic receptors, named keramadine 1 from the sea sponge.

The methanolic extract of the sea sponge <u>Agelas</u> sp., which was collected at Kerama Rettö, Okinawa, was suspended in water and extracted with n-butanol. The n-butanol soluble portion of the extract was chromatographed on columns of silica gel (chloroform-methanol 1:1, chloroform-n-butanol-acetic acid-water 25:60:16:10, and isoamylalcohol-acetic acid-water 30:13:10) and Sephadex LH-20 (methanol and chloroform-methanol 1:1) by monitoring the antagonistic activity on serotonergic receptors in the isolated aorta to give an active fraction. The fraction was chromatographed on a Develosil ODS column by using 6:4 water-methanol containing 0.2% trifluoroacetic acid as eluant to obtain an active substance, keramadine <u>1</u>, as colorless powder (mp. 183-187°C, 0.0014% yield from the fresh sponge).

The field desorption mass spectrum of <u>1</u> showed intense M+H ions at m/z 324 and 326, indicating that <u>1</u> is a monobromo compound ( $C_{12}H_{14}N_50Br$ ). The i.r. spectrum of <u>1</u> (KBr) showed an amide carbonyl absorption at 1680cm<sup>-1</sup>. A detailed analysis of the <sup>1</sup>H n.m.r. spectrum of <u>1</u> revealed a partial structure

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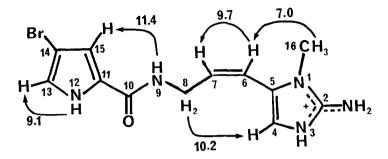


Fig. 1. Chemical structure of keramadine <u>1</u> and n.O.e.s (%), H(irradiated)  $\longrightarrow$  H(enhanced).

Position	$_{\delta\mathrm{H}}\mathbf{a}$	H, m, <u>J</u> (Hz)	δCp	mc	$\underline{J}_{H-C}$ (Hz)
2			146.9	s	
3	11.96	brs			
4	7.02	S	112.0		
5			123.7	d	200
б	6.20	AB, d, 11	133.3	d	160
7	5.81	$ABX_2$ , dt, 11 and 5.6	113.8	d	158
8	4.01	t, 5.6	38.6	t	
9	8.22	brt, 5.6			
10			159.6	S	
11			126.7	S	
12	11.58	brs			
13	6.92	dd, 2.9, 1.5	121.3	d	185
14			95.0	s	
15	6.80	dd, 2.9, 1.5	111.7	d	177
16	3.38	s	29.2	q	142
N(2)	7.59	brs			

Table 1.  $^{13}$ C n.m.r. (22.5 MHz) and  $^{1}$ H n.m.r. (90MHz) spectral data for keramadine <u>1</u> in DMSO-d<sub>6</sub>.

a;  $\delta$  in ppm, 70°C. b;  $\delta$  in ppm and assignments are based on single-frequency decoupling experiments. c; Multiplicity in the off-resonance decoupled spectrum.

-CO-NH-CH<sub>2</sub>-CH=CH-(cis) [ & 8.22 (9-H,exchangeable, brt, J=5.6 Hz), 4.01 (8-H. t, J=5.6 Hz), 5.81 (7-H, ABX, J=11 and 5.6 Hz) and 6.20 (6-H, d. J=11 Hz)]. The large H-C coupling constants (Table 1) indicated the presence of nitrogen containing heteroaromatic rings $^{4,5}$ , in agreement with the u.v. absorption,  $\lambda \max(MeOH)$  269 nm ( $\varepsilon$  21400). The <sup>1</sup>H n.m.r. spectrum of 1 contained an exchangeable signal at 611.58 (12-H) and signals for two aromatic protons at δ6.80 (15-H, dd, J<sub>13-15</sub>=1.5 Hz, J<sub>12-15</sub>=2.9 Hz) and 6.92 (13-H, dd, J<sub>15-13</sub>=1.5 Hz,  $J_{1,2-1,3}=2.9$  Hz), indicating the existence of a 2,4-disubstituted pyrrole ring. The pyrrole ring was further suggested by a positive color test with Ehrlich reagent. The substitution pattern of the pyrrole ring was determined on the basis of H-C coupling constants of C-13 (185 Hz) and C-15  $(177 \text{ Hz})^4$  and chemical sifts of C-11 ( $\delta$ 126.7) and C-14 ( $\delta$ 95.0)<sup>4,6</sup>. An N-methyl-2aminoimidazole unit was suggested by the signals at  $\delta 146.9$  (C-2), 112.0 (C-4), 123.7 (C-5) and 29.2 (C-16) and a large H-C coupling constant of C-4 (200  $Hz)^{5,6,9}$  The substitution position of the methyl group and the relation among these partial structures were elucidated by n.O.e. experiments (Fig. 1).

Keramadine <u>1</u> appears to be closely related biogenetically to brominecontaining alkaloids oroidin<sup>7,8</sup> and sceptrin<sup>9</sup> which have been isolated from sea sponges of the same genus <u>Agelas</u> as antimicrobial substances. However, the configuration of double bond at 6 position in <u>1</u> is the reverse of those in these compounds. In addition, in the isolated rabbit aorta the contractile response to serotonine  $(10^{-6}M)$  was abolished by <u>1</u>  $(1.5 \times 10^{-5}M)$ , whereas the responses to potassium chloride  $(4 \times 10^{-2}M)$  and norepinephrine  $(10^{-7}M)$  were not affected by <u>1</u>.

Acknowledgement: The authors gratefully acknowledge Dr. T. Hoshino (Mukaishima Marine Biological Station, Hiroshima University) for his kind identification of the sea sponge, Prof. T. Miyazawa and Dr. Higashijima (Department of Biophysic and Biochemistry, Faculty of Science, University of Tokyo for the n.O.e. measurements, Mr. Z. Nagahama for his skillful assistance in collecting the sea sponge, and Miss R. Abe of this institute for her excellent technical assistance.

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(Received in Japan 24 February 1984)